

**NITROSATION REACTIONS AND THE CHEMISTRY OF
NITRIC OXIDE**

Noelle Craig Borrell

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D.L.H. Williams (Author of Nitrosation Reactions and the Chemistry of Nitric Oxide)

Nitrosation Reactions and the Chemistry of Nitric Oxide By D. L. H. Williams (University of Durham, U.K.). Elsevier BV: Amsterdam. xii +.

Biological nitric oxide signalling: chemistry and terminology

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Editorial Reviews. Book Description. Provides an update on previously published literature in the field of nitric oxide chemistry.

In particular the chemistry of the products of thiol nitrosation, S-nitrosothiols, has been Since the discovery of the important biological properties of nitric oxide.

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Bsame as A for human and bovine serum albumin. Lipid peroxidation can also be initiated indirectly by oxidation of metal centres.

Such effects are not likely due to NO itself since it is such a poor oxidant one- Nitrosylation of ferric Fe III complexes is generally considered to be reversible, due to lower binding affinities compared to the corresponding ferrous analogues see Lim et al. Gunaydin and Houk, Rate predicts that 1 at high thiol concentrations the reaction rates should be independent of thiol concentration and 2 at low thiol concentrations the reaction rates should approach first order dependence on thiol concentration.

We have already provided a kinetic argument against this possibility. It is w extended periods of use, KOH pellets lost their effectiveness, probably due to surface accumulation of nitrite. Bell, London, p.